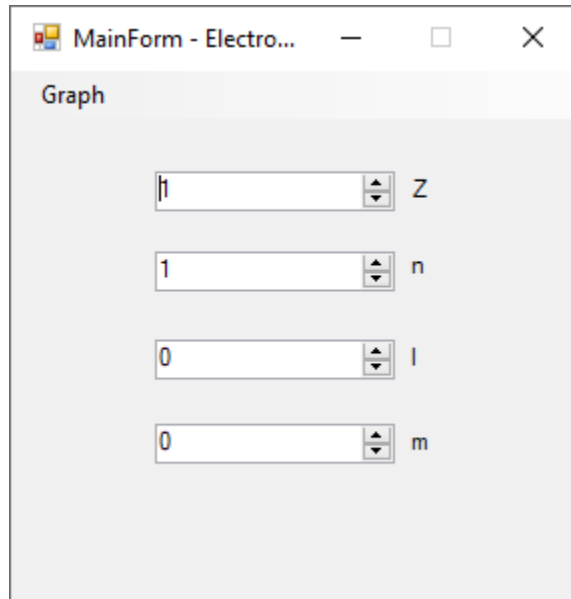
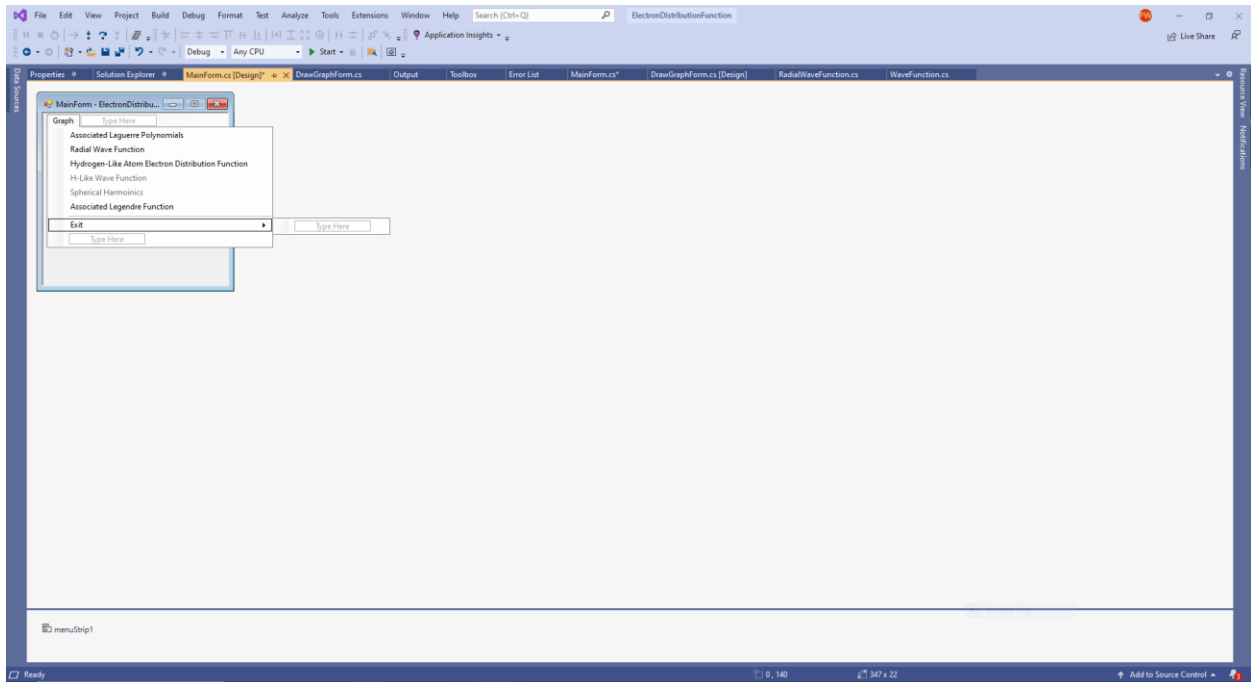


Electron Probability Distribution Function Etc. C# Application Designed and Implemented by James Pate Williams, Jr. © December 2023 All Applicable Rights Reserved

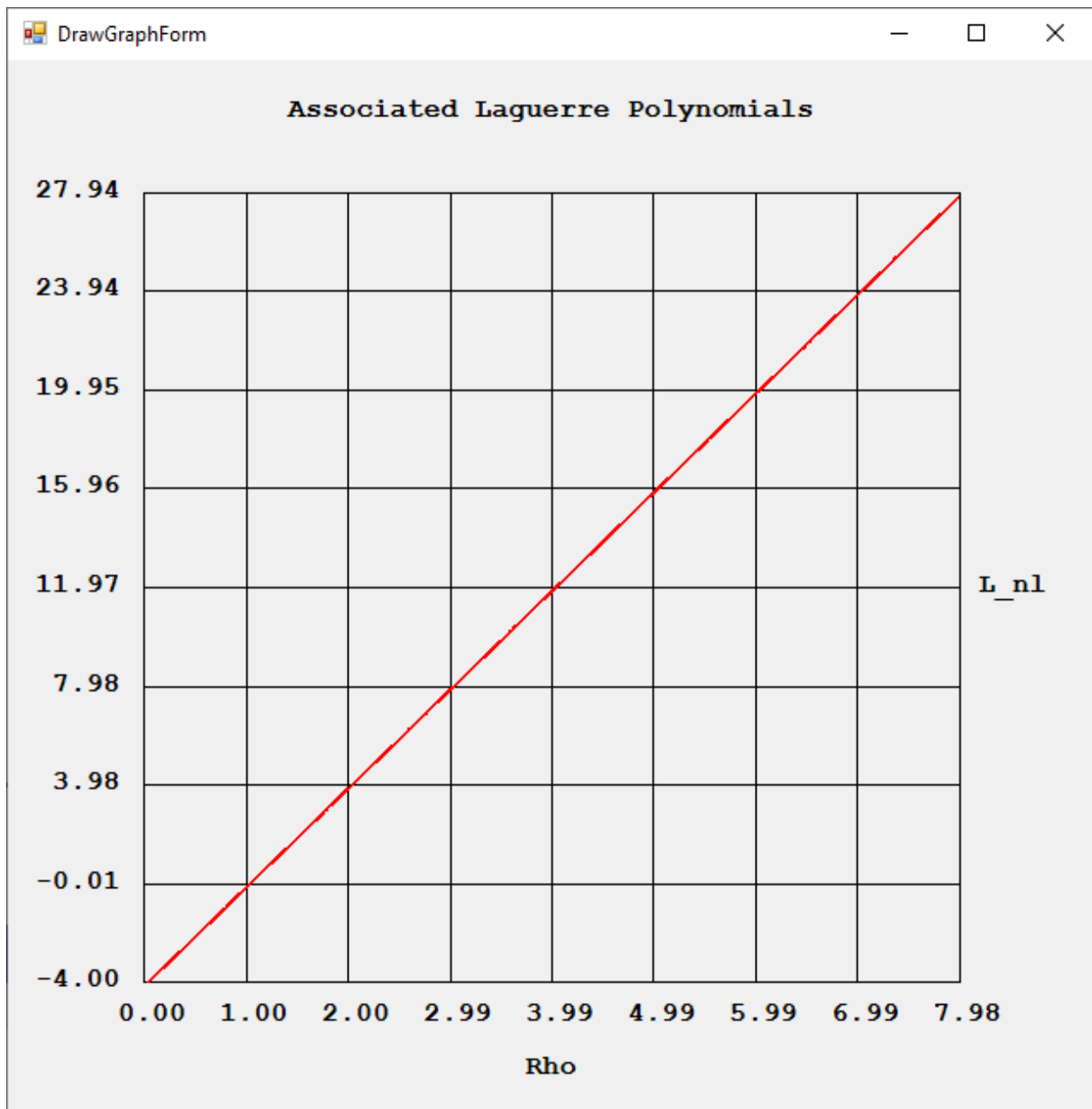
The opening menu data picker form:



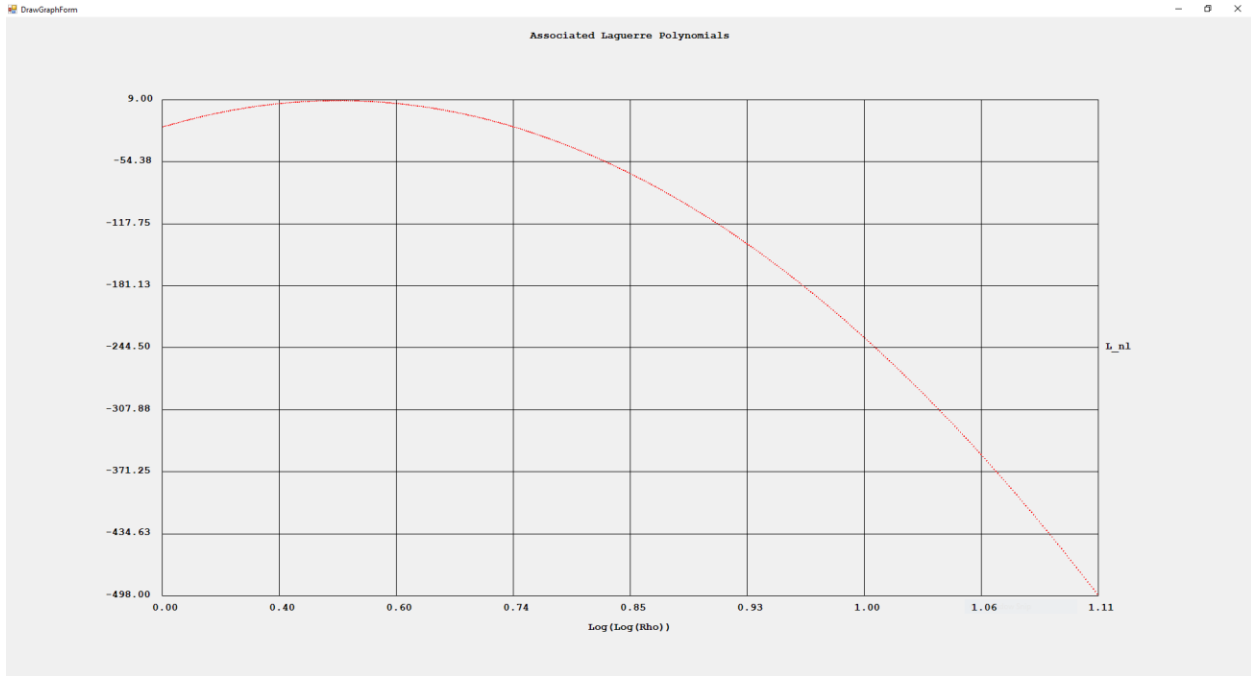
Z is the atomic number beginning at 1 for Hydrogen. n is the principal quantum number, l is the azimuthal quantum number, and m is the magnetic quantum number. The Graph menu is shown below:



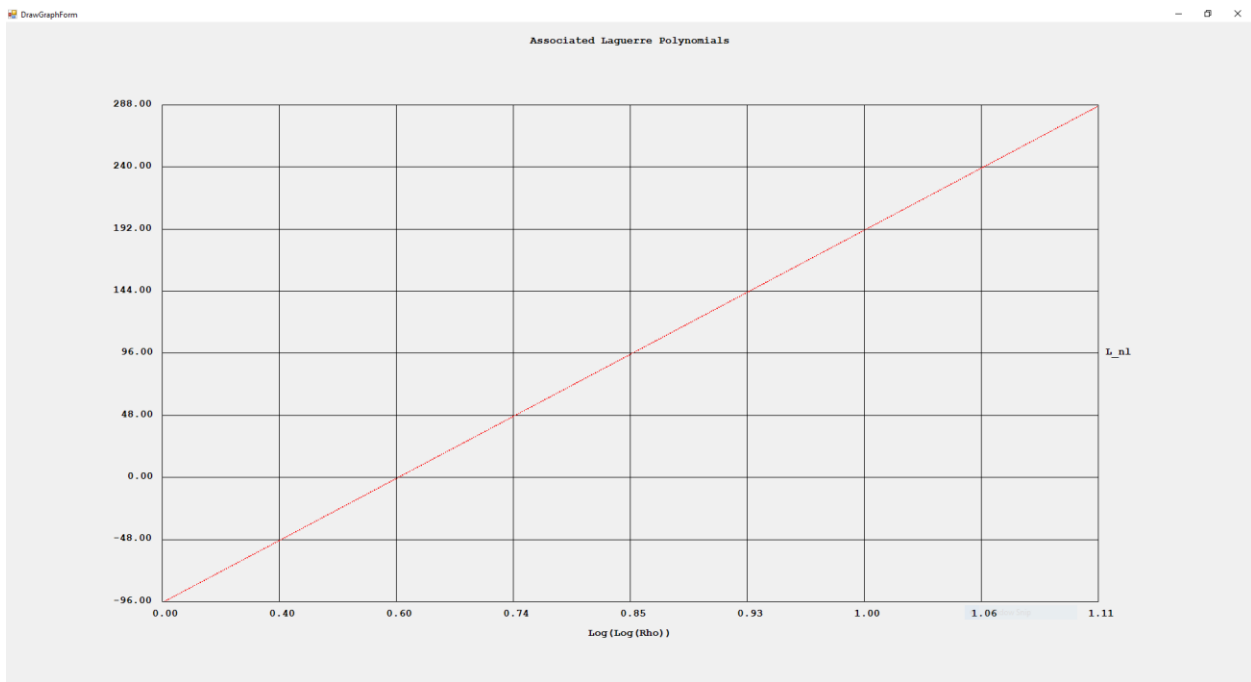
We start off with the Associated Laguerre Polynomials. For $Z = 2$, $n = 2$, $l = 0$ we have:



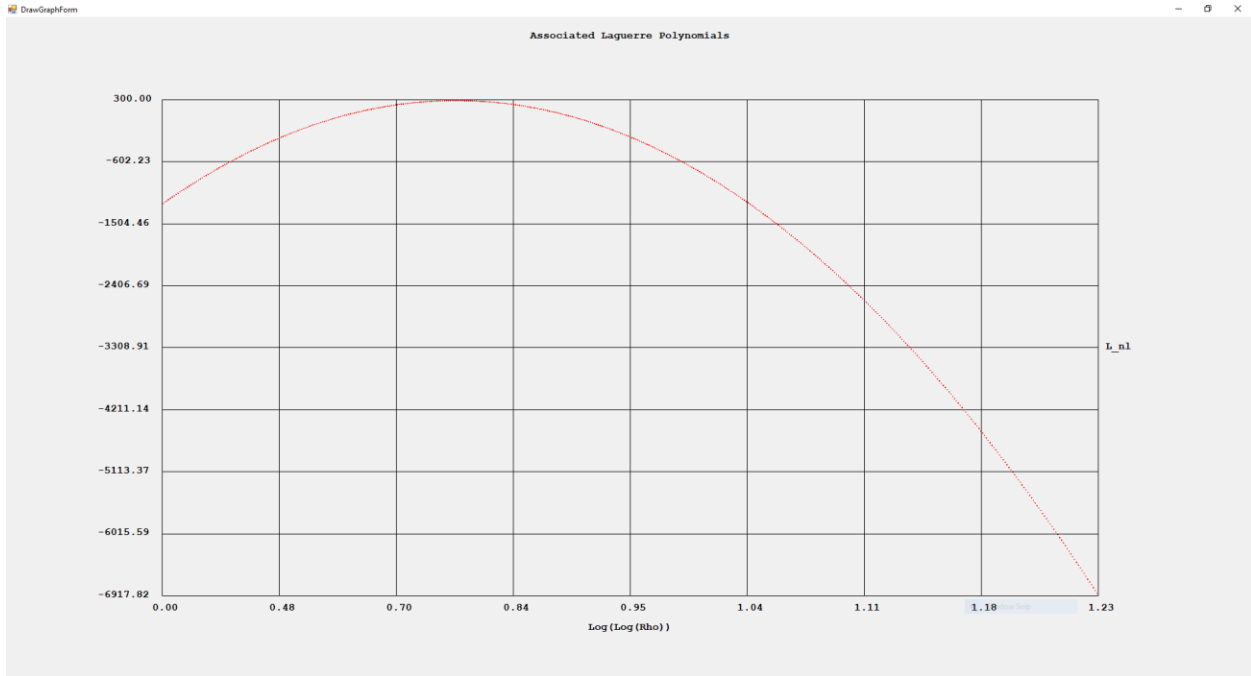
For $Z = 2, n = 3, l = 0$ yields:



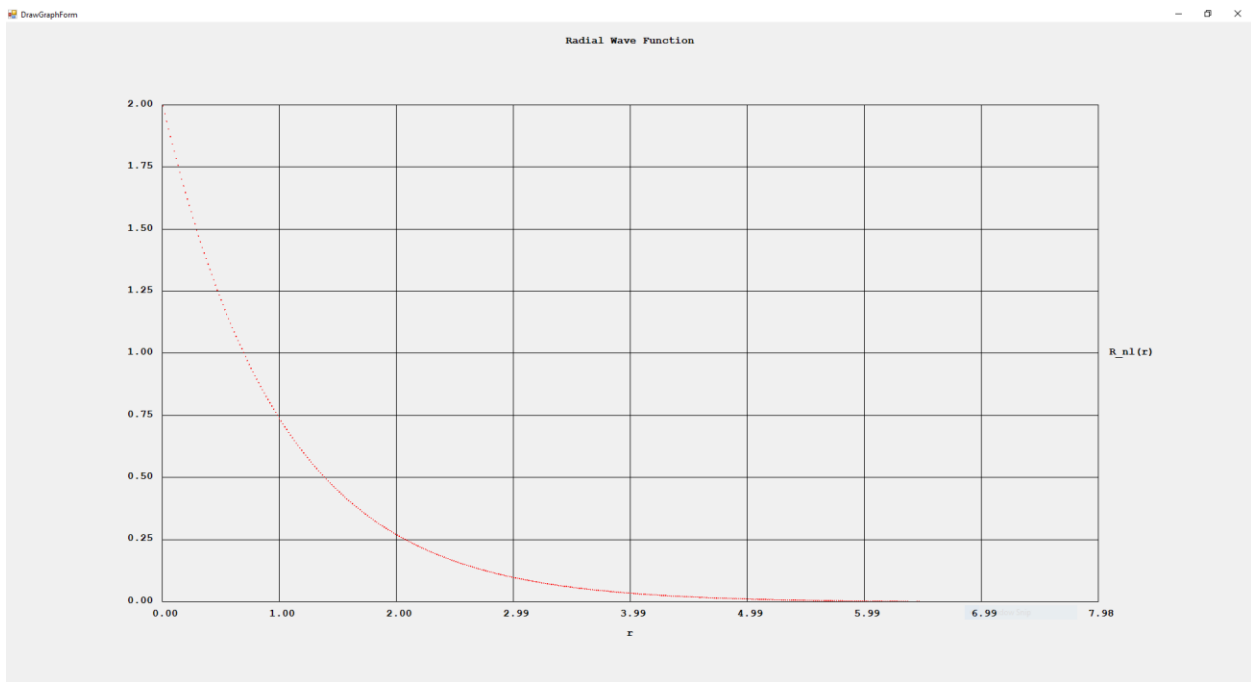
Let $Z = 2, n = 3, l = 1$:



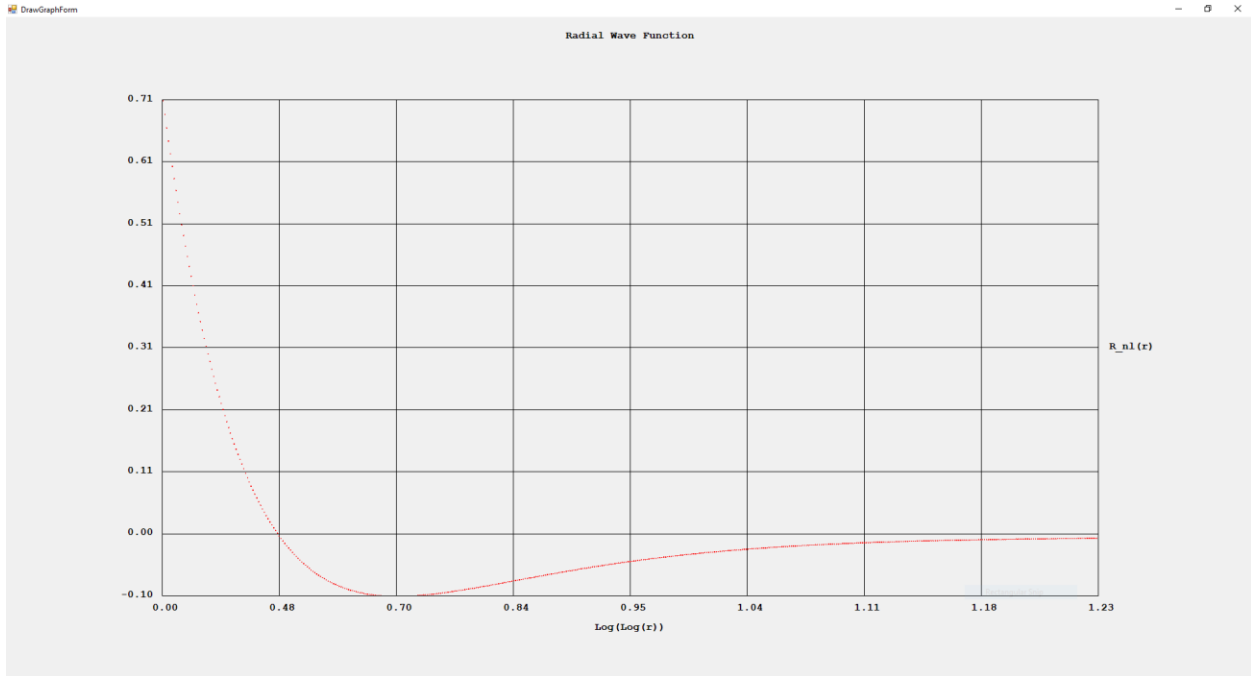
$Z = 2, n = 4, l = 1$:



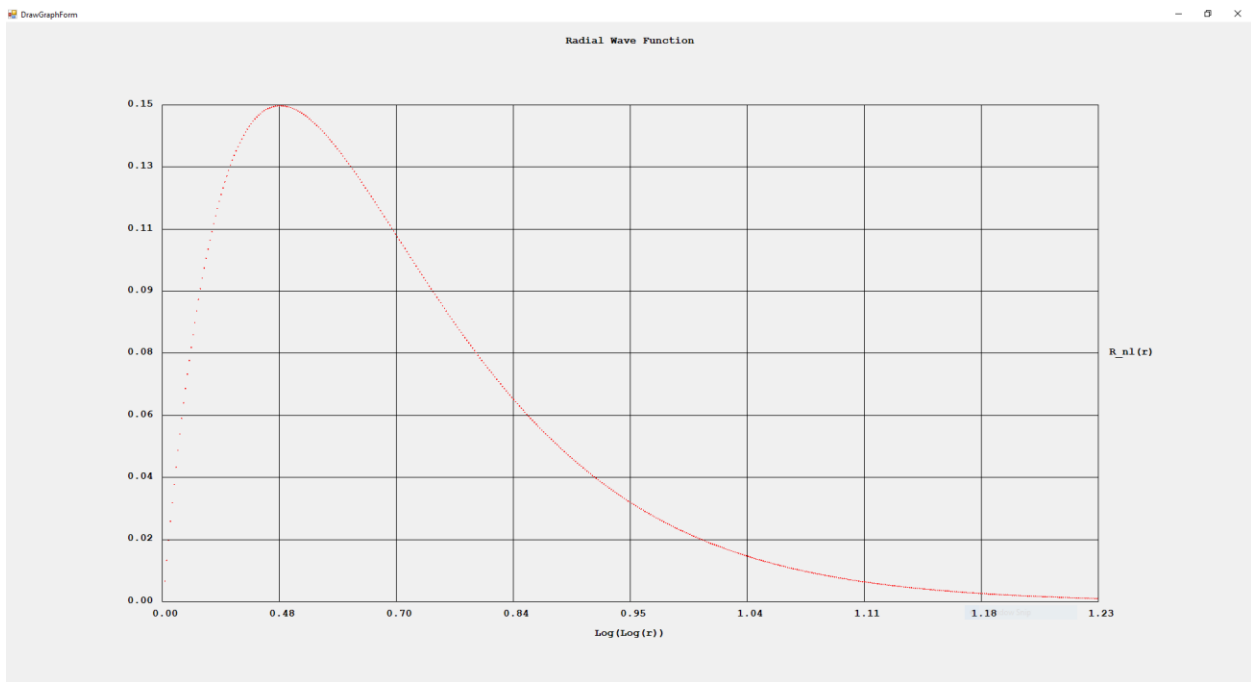
Results for the radial wavefunction: $Z = 1, n = 1, l = 0$:



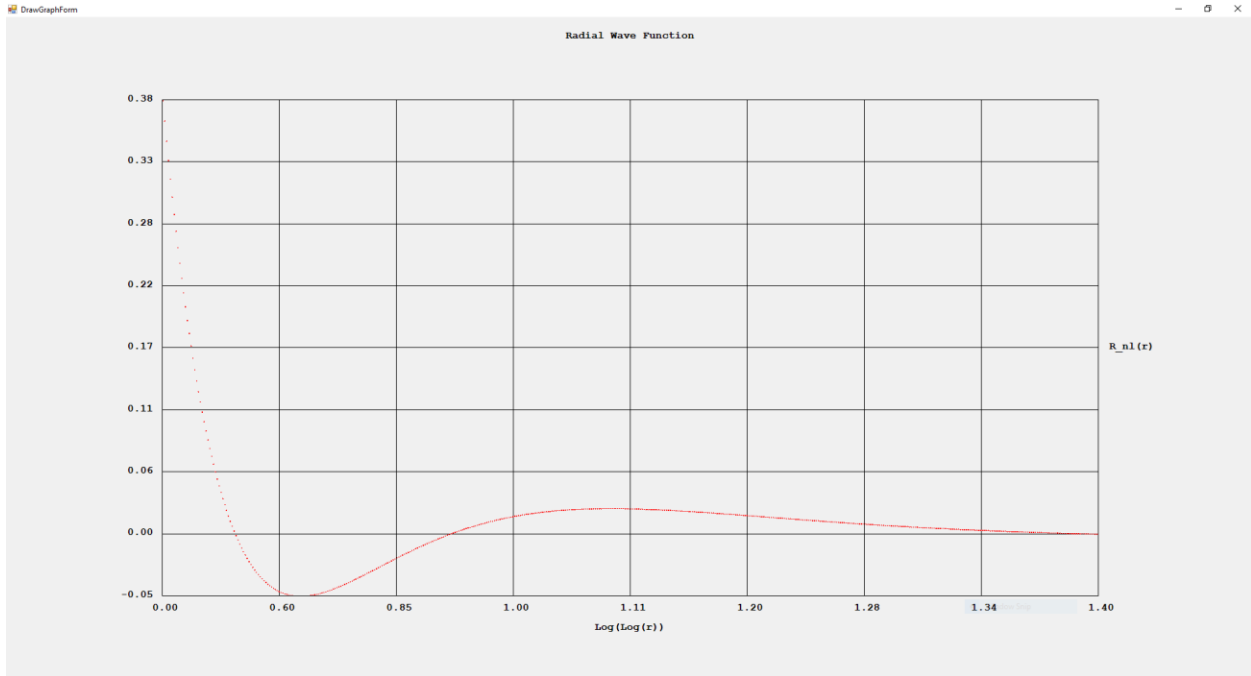
$Z = 1, n = 2, l = 0$:



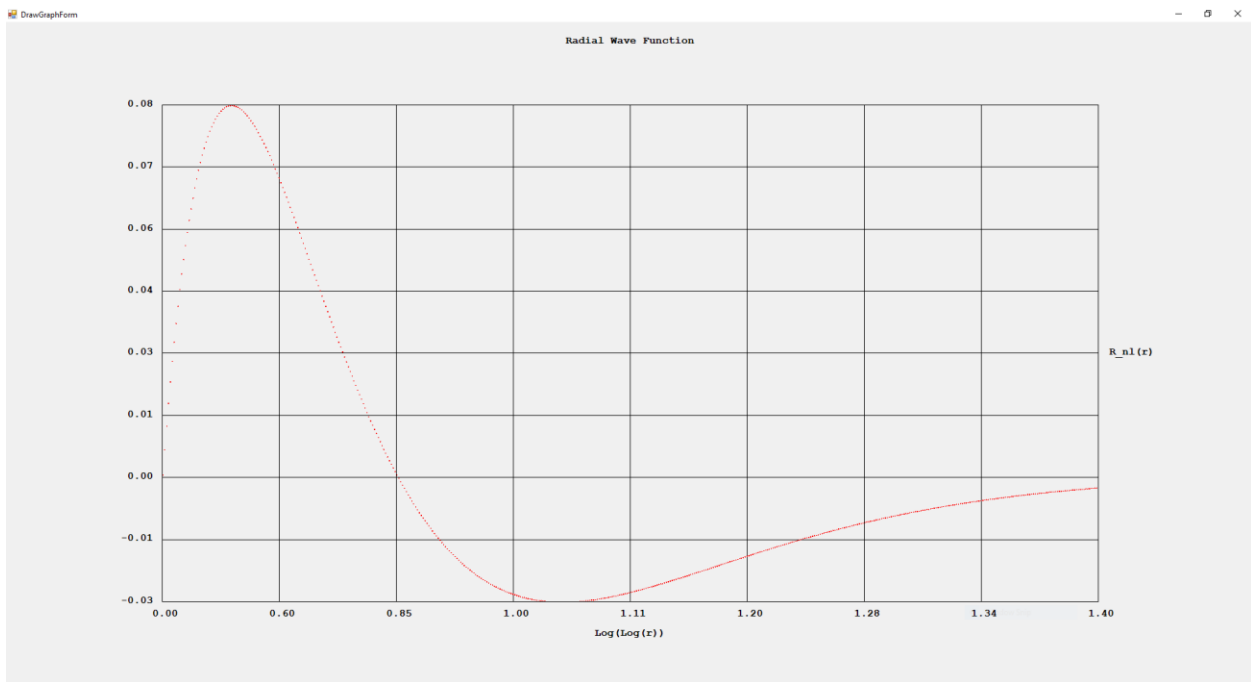
$Z = 1, n = 2, l = 1:$



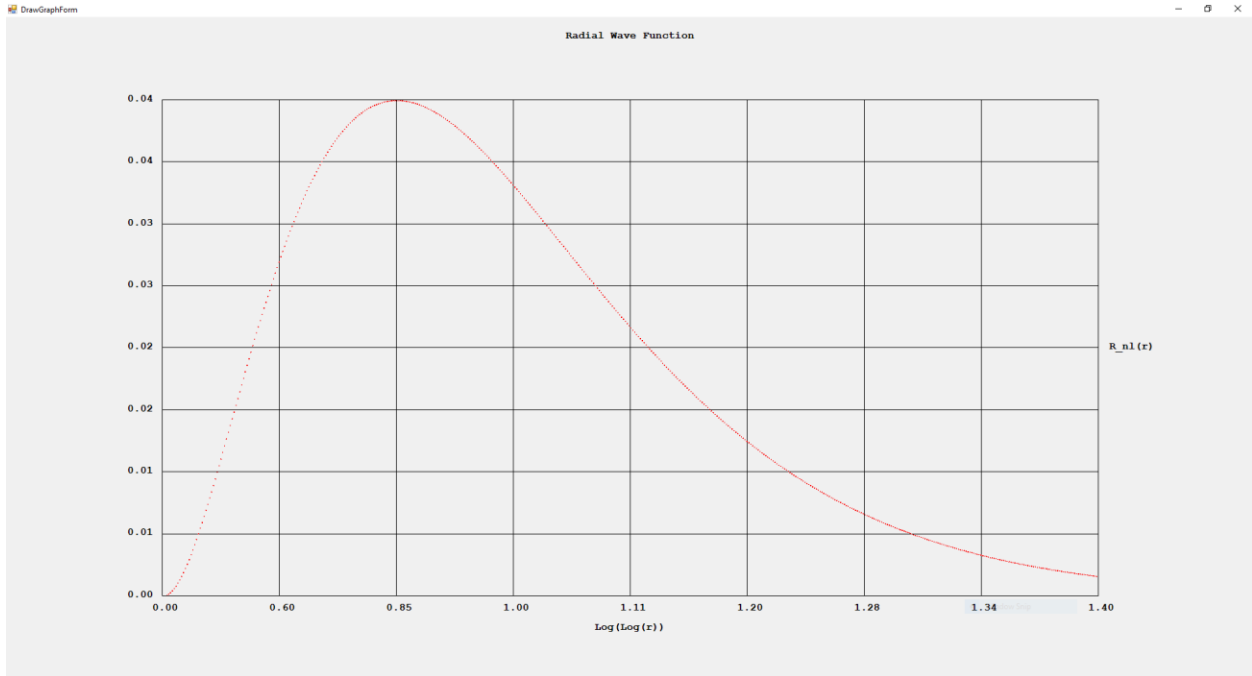
$Z = 1, n = 3, l = 0:$



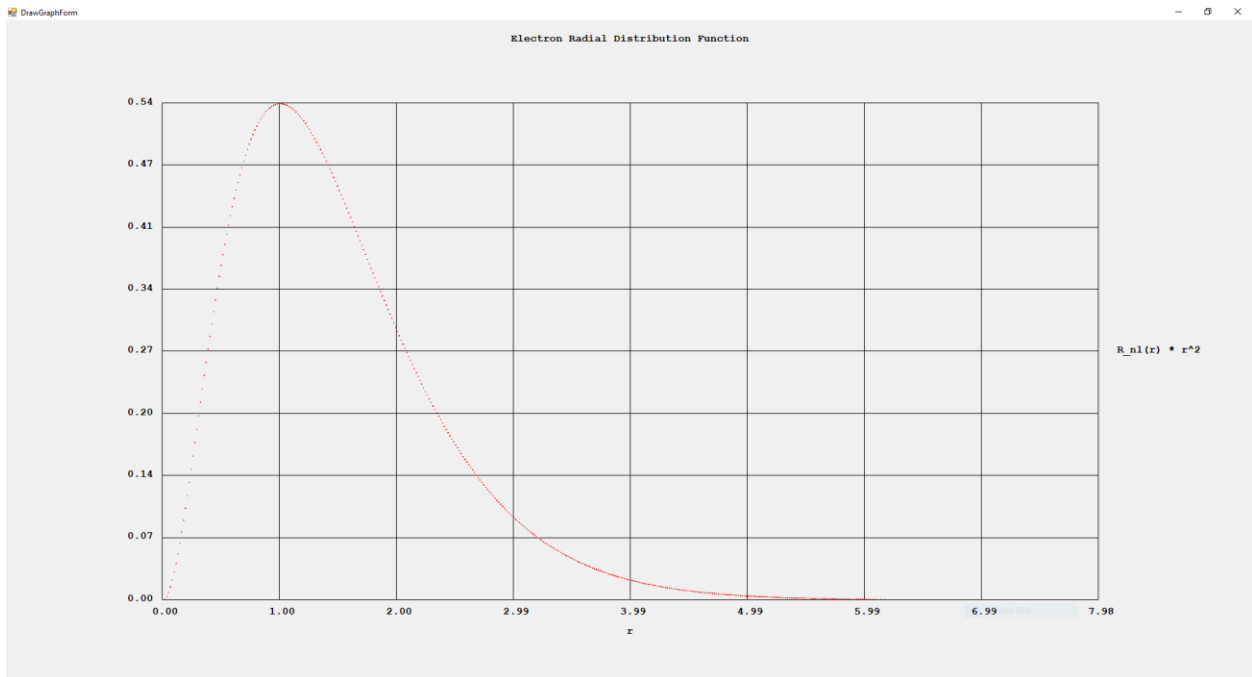
$Z = 1, n = 3, l = 1:$



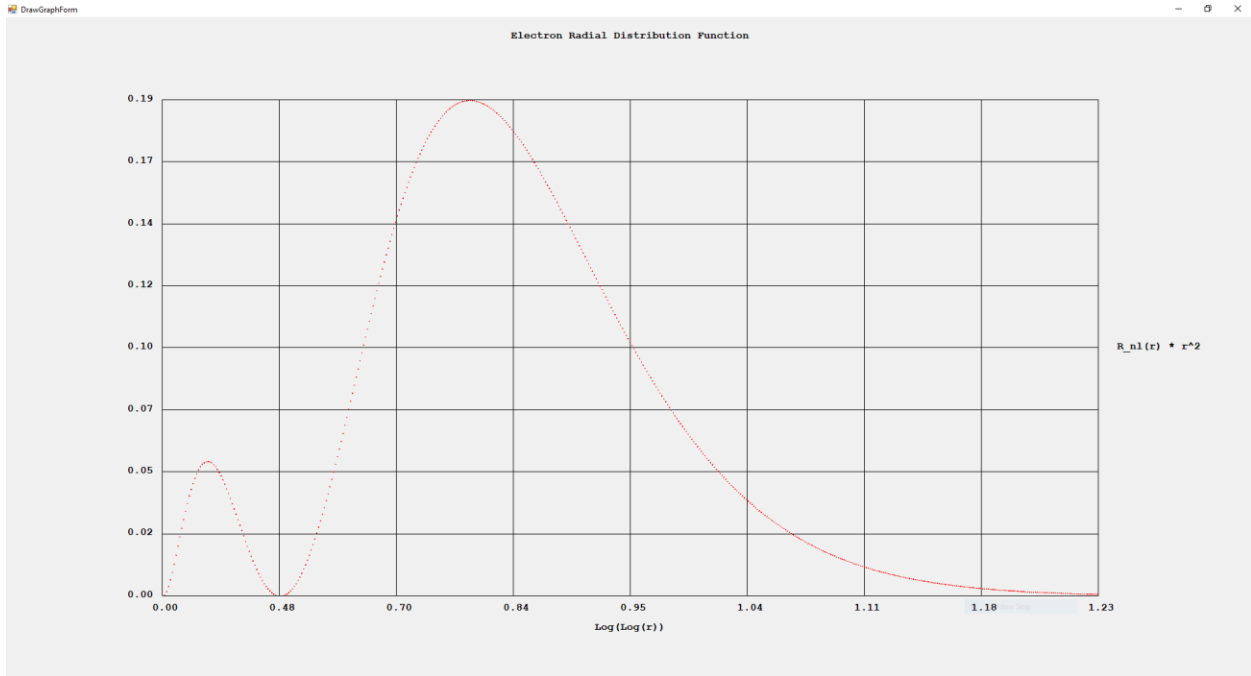
$Z = 1, n = 3, l = 2:$



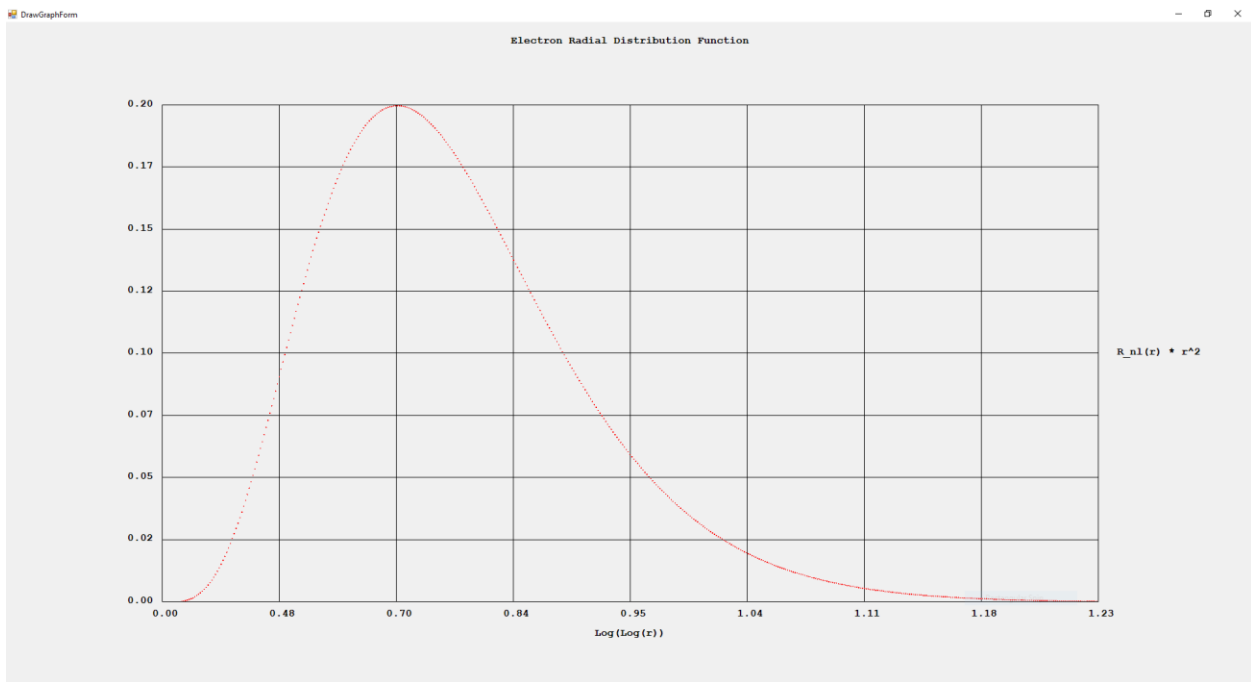
Now we move onto the hydrogen-like atom electron probability distribution function (PDF, not to be confused with a Portable Document File (another PDF)). $Z = 1, n = 1, l = 0$:



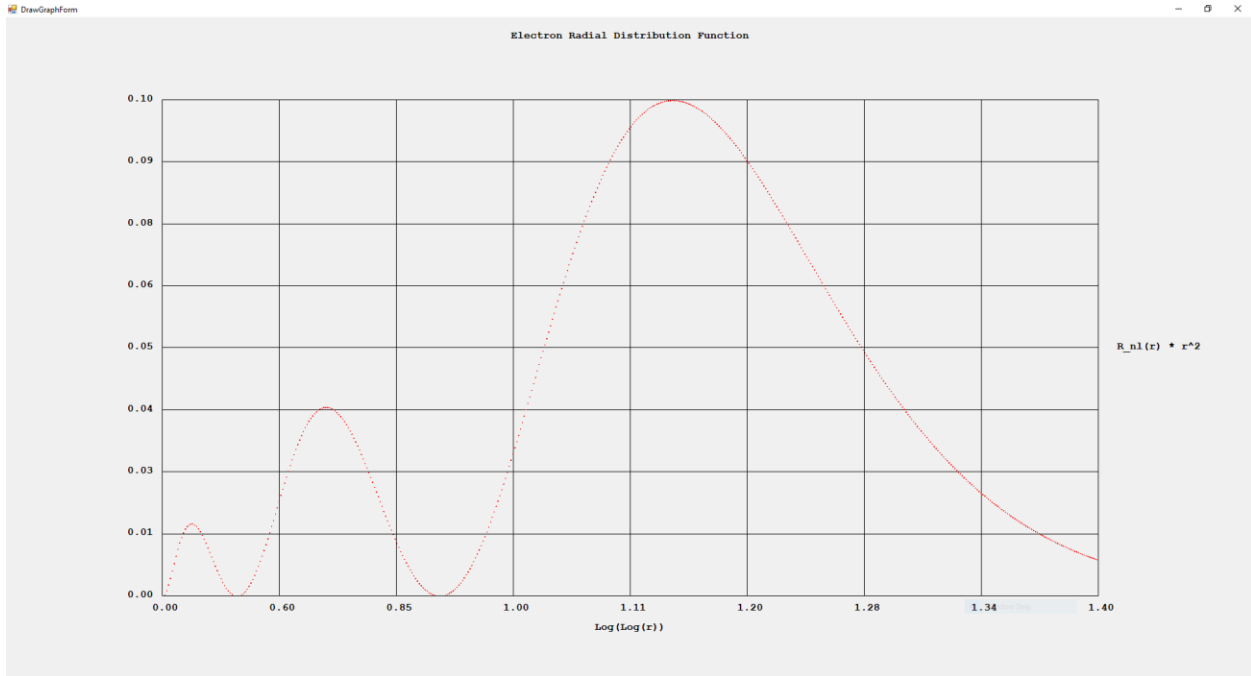
$Z = 1, n = 2, l = 0$:



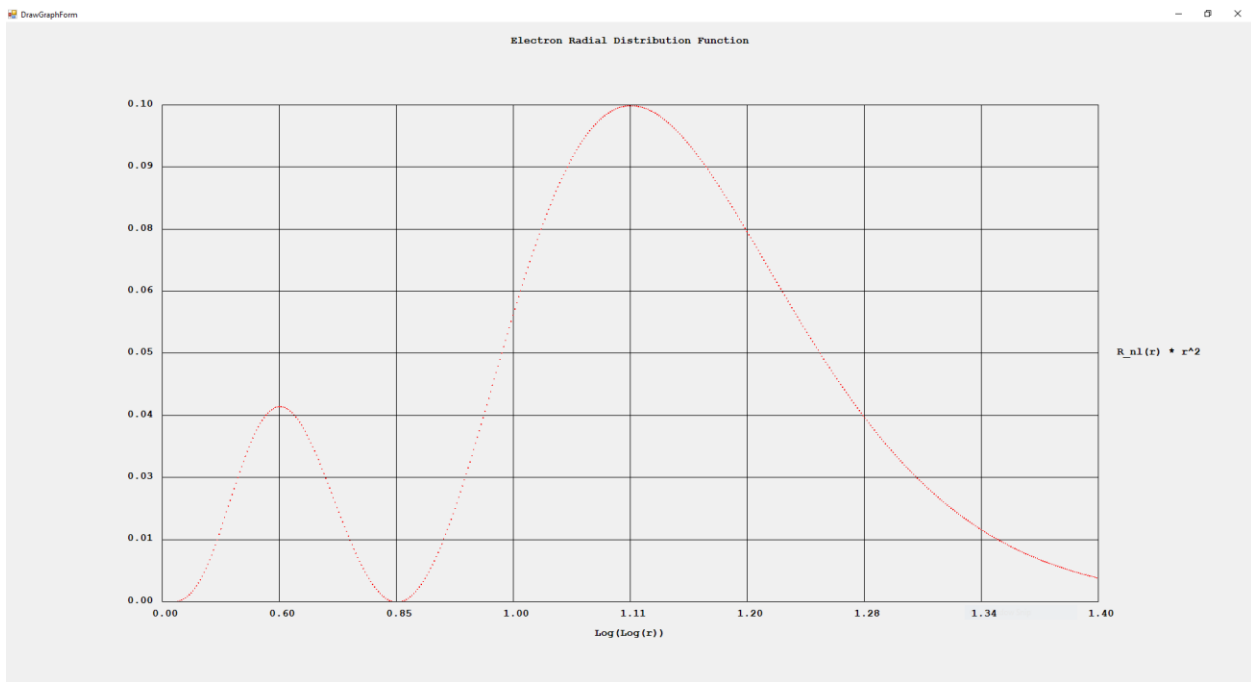
$Z = 1, n = 2, l = 1:$



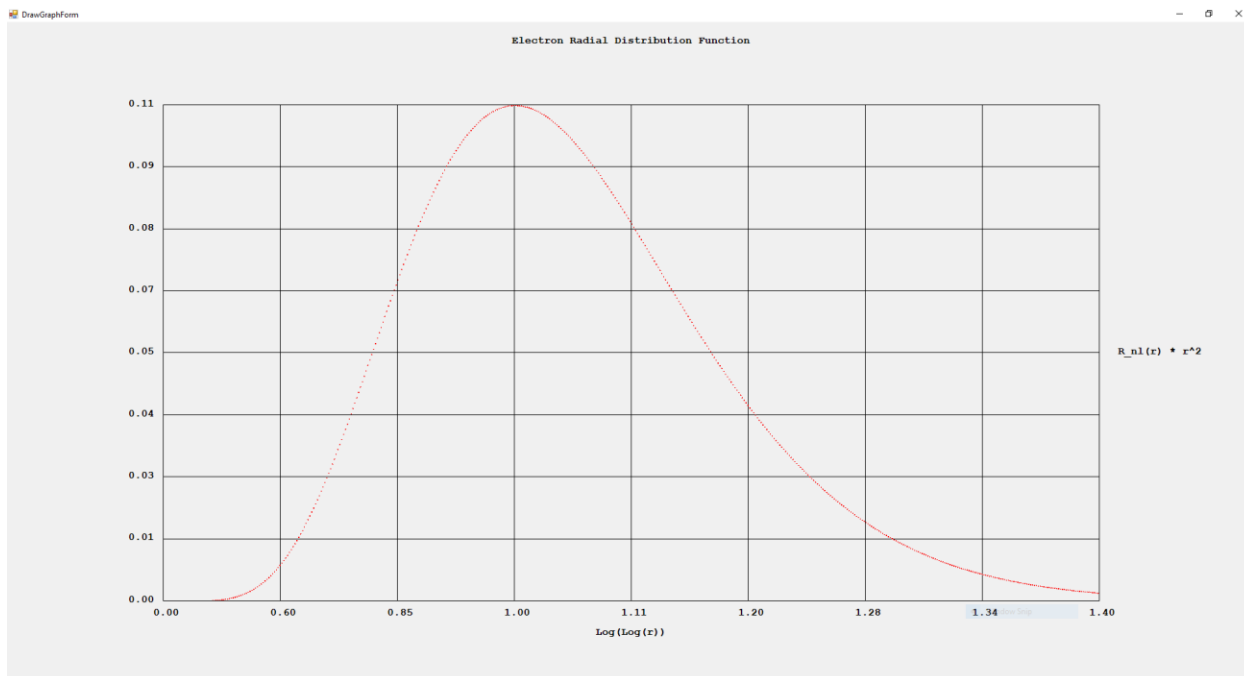
$Z = 1, n = 3, l = 0:$



$Z = 1, n = 3, l = 1:$



$Z = 1, n = 3, l = 2:$



References

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